Instructor: Dr. Valeria Martinovic Lecture: M,T,W, R 2:30 - 3:45 - in S32 **Lab:** M,T,W,R 11:30 - 2:20 SC2204

CRN: 00213 **Sections**: 02

Contact Information:

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Regarding the course:

This course syllabus is a contract. Please read it carefully and completely in its entirety before asking me any questions regarding the course schedule, content, requirements, grading, etc. You are expected to adhere to the De Anza College Student Code of Conduct Administrative Policy 5510 at all times.

The course is divided into a lecture period and a lab period. One registration code automatically enrolls you in both periods. At De Anza College, the lab and lecture cannot be taken as separate courses under any circumstances. Once you are enrolled you may not switch lab lecture or lab periods whether on a temporary or on-going basis.

Course Description

Chemistry 1B is the second quarter of a year-long introduction to the principles of general chemistry, and a direct continuation from Chemistry 1A. Chemistry 1B will constitute an investigation of intermolecular forces and their effects on chemical and physical properties, investigation of reversible reactions from the standpoints of kinetics, thermodynamics, and equilibrium, as well as investigation and application of gas laws and kinetic molecular theory.

Prerequisites

Chemistry 1A with a grade of C or better. EWRT 211 and READ 211 (or LART 211), or ESL 272 and 273.

Hours

Five hours lecture and twelve hours laboratory will be spent in class. The summer schedule is condensed, and we will cover two-weeks' worth of material every four days. You should expect to spend an additional 40 hours a week studying and working on class assignments.

Attendance Policy

Your punctual attendance is expected at all lecture and laboratory sections of the course. If you will have to miss class due to a verifiable emergency, let me know by e-mail as soon as possible. Exam and quiz dates may change. If you already know you will need to miss a day due to any other commitments, you should not take this class.

Textbook and Materials

1. Chemistry: The Molecular Nature of Matter and Change, 8th edition by Silberberg/Amateis. You are strongly encouraged to purchase this textbook from the De Anza College bookstore.

Dr. Martinovic

- 2. A scientific calculator. Phones and graphing calculators may not be used on exams or quizzes. I recommend the TI-30XS calculator which is available from multiple retailers.
- 3. The Chemistry 1B laboratory manual, available online at the De Anza Chemistry Department webpage. All lab procedures and pre-lab assignments will be posted on the course Canvas webpage.
- 4. A dedicated laboratory notebook. The lab notebook should be a bound composition type notebook. Notebooks with spiral binding and perforated pages are not permitted. All pages must be numbered.
- 5. Approved laboratory safety goggles (not safety glasses), available from the De Anza College Bookstore.
- 6. Disposable latex or nitrile gloves (recommended).
- 7. Stapler and staples.
- 8. Access to a printer.

Resources

- 1. Math, Sciences, and Technology Resource Center (MSTRC) Tutoring. The MSTRC offers tutoring for the Chemistry 1 sequence and is located in room S43 in the S-quad. Their website is: https://www.deanza.edu/studentsuccess/mstrc/
- 2. Disability Support Programs Services. The mission of DSPS is to ensure access to the college's curriculum, facilities, and programs. In particular, DSPS can help you get extended time on examinations. Their website is: https://www.deanza.edu/dsps/

Study Tips

- 1. Complete the assigned reading before coming to class. Review 1A topics that are unfamiliar.
- 2. Take handwritten notes during class and review your notes regularly. Write down any questions you have and bring them to class or office hours.
- 3. Do a little bit every day. Do not leave homework assignments until the last minute.
- 4. Join a study group. Work on problem sets together. The best way to learn the material is to teach it to somebody else.
- 5. If you feel that you are a poor test-taken, complete and turn in all other assignments on time in order to pass the class.
- 6. Take care of yourself! Stay well-rested and drink water.

Grading Breakdown and Grade Scale

To succeed in this course, you will need to exhibit consistent and sustained effort throughout the quarter. This will be demonstrated through homework assignments, laboratory preparation and data analysis, and examinations.

Exam 1	100
Exam 2	100
Final Exam	200
Participation in class and in lab	100
Laboratory Exam	100
Laboratory work	10 points/lab report

A+≥97%	C+ ≥ 75%
A ≥92%	C ≥67%
$B+\geq 86\%$	D ≥ 58%
B ≥ 82%	D-≥55%
B−≥79%	F < 55%

If your average in either the lab or lecture portion of the course is less than 55%, you will not pass the class. A+ grades will be given to students who demonstrate excellence in the following three areas: lecture, lab and class participation.

Lecture

Your attendance and active participation is expected at every lecture period. Due to the high number of students wishing to enroll in the course, any unjustified absences during the first two days of class will result in you being dropped from the course. Absences may be excused in case of a verified emergency (e.g. doctor's note or police report). If you know that you will not be able to attend lecture for any reason, let me know by email right away. The lecture participation grade (5%) must be earned through active engagement during class. Late arrivals and early departures are distracting for the whole class (and me!), so arrive on time and stay for the whole class period. I strongly encourage taking your own notes in lecture. Computers are not necessary during lecture. Do not use your computers for noncourse related activities during lecture. Put your phone on silent or Do Not Disturb while you are in class. If you must take a phone call in case of emergency, quietly leave the room before answering the phone.

Exams

There will be two midterm exams, each worth 100 points. Early and late exams will not be administered, and missing an exam will result in a zero without documented proof of a medical or legal emergency (e.g. hospitalization or car crash). If you need any accommodations for exams, DSPS will be able to notify me through Clockwork.

Exams will consist of short answer questions with the opportunity for partial credit. You must show your work in order to receive credit for any answer. I am more interested in how you think about a problem than your final answer. You will be asked to demonstrate your conceptual understanding of the material and apply those concepts in an algebraic context and solve quantitative problems.

Final

The final exam will be cumulative. The final exam will be administered on Thursday, August 9th from 8:20 am - 10:20 am. If you cannot take the final at this time, you should not enroll in the class. The final will not be administered at an alternative time under any circumstances. You must take the final to pass the course.

Lab

Chemistry is an experimental science, and the laboratory is a major component of the course. De Anza College does not offer make-up labs, and you must attend the laboratory section that you are registered for to complete the required labs. Everyone gets one excused absence with no grade penalty. To receive this excused absence, you must contact your instructor ASAP to explain your absence from lab. After 24 hours, you will not able to excuse this absence and you will receive a zero for all associated assignments. A second absence, regardless of the circumstances of your first absence, will result in a zero for the lab and all associated assignments for that lab. After a third lab absence, you will

automatically receive an "F" in the course.

Your timely attendance is expected at every lab. The beginning of each lab period is reserved for lab lecture. The lab lecture is a required component of the laboratory section and will include essential safety information. If you miss lab lecture, you will not be permitted to complete that lab and you will receive a zero for all related assignments (e.g. Pre-lab, lab data, and lab analysis).

You must clean up your work area before leaving each lab. Failure to do so will result in a loss of points for that lab. Before you leave lab, check-out with me. You will not receive credit for the lab unless I have signed your data.

Pre-lab

Pre-labs must be prepared in your laboratory notebook before the start of your laboratory section. I will check your pre-lab at the start of class.

Lab Data

Each wet-lab day is worth a total of 10 points. Data collected during the lab period must be recorded in your laboratory notebook. If you write any data on a separate sheet of paper or the lab manual, you will lose points for that lab. You may recopy your data into a clean table in your lab notebook later if you wish. Before you leave lab for the day, have me check off on your data and calculations in your

Lecture Exam 1: July 11 Lecture Exam 2: July 23 Lab Final: August 7

Lecture Final: August 8 (at 11:30am)

lab notebook for the available points.

Lab Final

There will be one lab exam in this course. The lab final will be an open lab-notebook exam, and you may refer to any information that is handwritten in your lab notebook. Extra pages (either printed or handwritten) may not be inserted. The final will cover material, calculations, and analysis related to your laboratory experiments.

Clean-up

Each student is required to sign up for one lab period in which they will be responsible for after-lab clean-up. This involves staying to end of lab, making sure the common lab areas and balance area is clean, the waste bottles are closed, etc. In addition, each student is responsible for cleaning their own materials and work area. All clean-up must begin by 2pm.

Important Dates:

The fist day of class: July 1, 2019

Independence Day Holiday: July 8, 2018

If you drop or withdraw from the course, you must check out of your lab locker by the last lab

Academic Integrity

Students are expected to adhere to the policy on academic integrity that is outlined in the De Anza College manual (https://www.deanza.edu/studenthandbook/academic-integrity.html). I expect all submitted work to represent your own understanding of the material. Cheating, copying, plagiarizing, etc. will not be tolerated, and the minimum consequence will be receiving a zero on that assignment. All laboratory data used in calculations and reported in lab reports must be collected by each student. Multiple instances of academic dishonesty may result in failing the course.

Copying any assignment from another student is cheating. If I see you copying an assignment, both students will receive a zero on that assignment.

Lecture and Lab Schedule

Chemistry 1B will cover material presented in chapters 5, 12, 16, 17, 18 and 20 of Silberberg. We will also review Chemistry 1A topics presented in chapters 10 and 6 throughout the quarter. Every effort will be made to keep to the lecture schedule below. If we fall significantly behind this schedule, the content of the exams will be adjusted to reflect the material that we covered in class. Exam and quiz dates by change at your instructor's discretion, so plan to attend all days.

	Monday	Tuesday	Wednesday	Thursday
Week 1 July 1– Lab	Syllabus Check-in	Molar volume Day 1	Molar volume Day 2	NO CLASS

Lecture	Ideal Gases Chapter 5	Ideal Gases Chapter 5	Intermolecular Forces Chapter 12	NO CLASS
Week 2 July 8 – 11 Lab	Vapor Pressure Day	Vapor Pressure Day 2	Green Salt Day1	Green Salt Day 2
Lecture	Intermolecular Forces Chapter 12	Phase Changes Chapter 12	Chemical Kinetics Chapter 16	Exam 1
Week 3 July 15 – 18 Lab	Green Salt Day 3	Green Salt Day 4	Iodine Clock Day 1	Iodine Clock Day 2
Lecture	Chemical Kinetics Chapter 16	Chemical Kinetics Chapter 16	Equilibrium Chapter 17	Equilibrium Chapter 17
Week 4 July 22 – 25 Lab	Iodine Clock Day 3	Iodine Clock Day 4	K _c of iron Day 1	K _c of iron Day 2
Lecture	Equilibrium Chapter 17	Exam 2	Acid-base chem Chapter 18	Acid-base chem Chapter 18
Week 5 July 29 – Aug 1 Lab	K _A of Weak Acid Day 1	pK _A of Indicator Day 1	pK _A of Indicator Day 2	calcium hydroxide Day 1
Lecture	Acid-base chem Chapter 18	Acid-base chem Chapter 18	Acid-Base chem Chapter 18	Thermodynamics Chapter 20
Week 6 Aug 5 – 8 Lab	calcium hydroxide Day 1	Check-out	Lab final	Lecture final 11:30 – 1:30 am
Lecture	Thermodynamics Chapter 20	Thermodynamics Chapter 20	Review	Have a great break!

From the American Chemical Society Safety In Academic Laboratories Guidelines, 7th Ed., the following mandatory minimum safety requirements must be followed by all students and be rigorously enforced by the entire Chemistry faculty:

- 1. Chemistry Department-approved safety goggles purchased from the De Anza College bookstore (NOT safety glasses) must be worn at all times once laboratory work begins, including when obtaining equipment from the stockroom or removing equipment from student drawers, and may not be removed until all laboratory work has ended and all glassware has been returned to student drawers.
- 2. Shoes that completely enclose the foot are to be worn at all times; NO sandals, open-toed, or open-topped shoes, or slippers, even with socks on, are to be worn in the lab

- 3. Shorts, cut-offs, skirts or pants exposing skin above the ankle, and sleeveless tops may not be worn in the lab: ankle-length clothing must be worn at all times
- 4. Hair reaching the top of the shoulders must be tied back securely
- 5. Loose clothing must be constrained
- 6. Wearing "...jewelry such as rings, bracelets, and wristwatches in the laboratory..." should be discouraged to prevent "...chemical seepage in between the jewelry and skin...".
- 7. Eating, drinking, or applying cosmetics in the laboratory is forbidden at ALL times, including during lab lecture
- 8. Use of electronic devices requiring headphones in the laboratory is prohibited at ALL times, including during lab lecture
- 9. Students are advised to inform their instructor about any pre-existing medical conditions, such as pregnancy, epilepsy, or diabetes that they have that might affect their performance.
- 10. Students are required to know the locations of the eyewash stations, emergency shower, and all exits
- 11. Students may not be in the lab without an instructor being present
- 12. Students not enrolled in the laboratory class may not be in the lab at any time after the first lab period of each quarter.
- 13. Except for soapy or clear rinse water from washing glassware, NO CHEMICALS MAY BE POURED INTO THE SINKS; all remaining chemicals from an experiment must be poured into the waste bottle provided.
- 14. Students are required to follow the De Anza College Code of Conduct at all times while in lab: "horseplay", yelling, offensive language, or any behavior that could startle or frighten another student is not allowed during lab;
- 15. Strongly recommended: Wear Nitrile gloves while performing lab work; wear a chemically resistant lab coat or lab apron; wear shoes made of leather or polymeric leather substitute.

By signing below			,
	First Name	Family Name	
_	vledge that my failure to ab	ee to abide by the laboratory safety rules lis ide by these rules will result in my being dro	
Signature		Date	

Student Learning Outcome(s):

- *Evaluate the principles of molecular kinetics.
- *Apply principles of chemical equilibrium to chemical reactions.
- *Apply the second and third laws of thermodynamics to chemical reactions.